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### TEST 5 Answers CHEM 11 Atomic Structure NAME: \_\_\_\_\_

An isotope of any element can be uniquely represented as  $(^A_Z X)$ , where X is the atomic symbol of the element. The isotope of carbon that has 6 neutrons is therefore  $(_{6}^{12} C)$ . The subscript

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indicating the atomic number is actually redundant because the atomic symbol already uniquely specifies Z.

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Atomic weight. The atomic mass unit (amu) is often used to express atomic weight. 1 amu is defined as 1/12 of the atomic mass of the most common isotope of carbon atom that has 6 protons ( $Z=6$ ) and six neutrons ( $N=6$ ).  $m_{\text{proton}} \approx m_{\text{neutron}} = 1.66 \times 10^{-24} \text{ g} = 1 \text{ amu}$ . The atomic mass of the  $^{12}\text{C}$  atom is 12 amu. The atomic weight of an element = weighted ...

### **Chapter Outline Review of Atomic Structure**

'Atomic Structure Review Answer Key studylib net April 13th, 2018 - The most abundant isotopes of hydrogen and Fill in the table below Symbol 79 34 15 Atomic number Se 2 34 Mass number 79

### **Atomic Structure Review Answers Table Isotope Symbol**

Atomic Theory and Structure Quiz. Multiple Choice Identify the choice that best completes the statement or answers the question. On the scantron sheet do the following: in the subject line, put My Favorite Class and in the period line put, The Smartest. \_\_\_\_ 1.

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